Ring-Closing Metathesis of Ene-Ynamide: Application to the Synthesis of Medium-Sized Cyclic Dienamide

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Summary: Ring-closing metathesis (RCM) of ene-ynamide, which could be applied to the synthesis of various heterocycles containing 7- and 8-membered rings, was investigated. Ene-ynamides are easily synthesized by the known method. When a toluene solution of ene-ynamide was stirred in the presence of a catalytic amount of second-generation ruthenium carbene complex 1 under an ethylene atmosphere, RCM proceeded smoothly to provide a heterocyclic compound having a diene moiety in good to high yield. A substituent of the ynamide moiety affected the yield of the cyclized product.

Keywords: metathesis; medium-sized heterocycles; ring-closing metathesis; synthesis; ynamide

Introduction

Ynamines and ynamides have been shown to be useful building blocks in organic synthesis. [1] The distinctive reactivity of ynamides, which have an electron-deficient π -orbital and stability relative to ynamines, is attractive for an organic chemist. Several kinds of transition metal-catalyzed reactions of ynamides have been reported in recent years. [2] We have studied the ringclosing metathesis of ene-ynamide, which can be applied to useful synthetic methods for a wide range of heterocycles. [3] Herein, we wish to report the synthesis of 7- and 8-membered rings using RCM of ene-ynamide.

Construction of 7-Membered Ring

At first, the construction of a benzazepine ring was examined. When a toluene solu-

tion of 2a was stirred with 10 mol% of second-generation Grubbs catalyst 1 at 80 °C for 1.5 h under an ethylene atmosphere, cyclized product 3a was obtained in 22% yield (Table 1, Entry 1).^[4] A corresponding reaction was carried out under an argon atmosphere to provide an improved yield of **3a** to 49% (Entry 2). We then tried to perform RCM of 2b, which was prepared by the introduction of an ethoxycarbonyl group on the alkyne. When a reaction of 2b was carried out in a similar manner, benzazepine derivative 3b was obtained in 69% yield (Entry 3). The yield could be enhanced to 84% by performing the reaction under an argon atmosphere (Entry 4).

Next, RCM of linear α,ω -ene-ynamides was examined. When a reaction of **4a** was carried out in toluene at 80 °C for 1 h, starting material **4a** was recovered in 53% yield (Table 2, Entry 1). Replacement of the solvent from toluene to CH_2Cl_2 lead to a progressive result, and **5a** was provided in 36% yield upon heating for 15 h, together with 38% of **4a** (Entry 2). [5] A similar result was obtained, when the reaction was conducted under an argon atmosphere or prolonged reaction time (Entry 3,4). In the case of **4b**, as a substrate having an



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Table 1.Synthesis of benzazepine derivatives.

Entry	Substrate	Atmosphere	1 (mol%)	Time (h)	3 (%) ^a
1	2a	CH ₂ =CH ₂	10	1.5	22
2	2a	Ar	5	0.5	49
3	2b	$CH_2 = CH_2$	10	3.5	69
4	2b	Ar	5	0.5	84

a) Isolated yield.

ethoxycarbonyl group on the alkyne, the yield of cyclized product $\bf 5b$ was only 5% when the reaction was carried out in toluene upon heating (Entry 5). However, a higher yield of $\bf 5b$ was obtained when the reaction was carried out in CH_2Cl_2 (entry 6).

Subsequently, the synthesis of benzodiazepine derivatives was examined. When a toluene solution of **6a** was exposed to a catalytic amount of **1** at 80 °C for 1 h, starting material **6a** was not consumed and an inseparable complex mixture, which

4b: R = CO₂Et

was composed of **6a**, cyclized product **7a**, an intermolecular metathesis product with ethylene, and undefined products, was obtained (Table 3, Entry 1). Replacement of the solvent and a longer reaction time lead to a similar result (Entry 2). However, we were very surprised to find that when **6b** was used as the substrate instead of **6a**, benzodiazepine **7b** was obtained in quantitative yield (Entry 3). [6] A reduction of the catalyst amount did not affect the yield of **7b** (Entry 4). RCM proceeded smoothly in CH₂Cl₂ as the solvent, although a longer

Table 2. Synthesis of azepine derivatives.

Entry	Substrate	Atmosphere	Solvent	Temp (°C)	Time (h)	5 (%) ^a	Recovery of 4 (%) ^a
1 ^b	4a	CH ₂ =CH ₂	Toluene	80	1	-	53°
2	4a	$CH_2 = CH_2$	CH_2Cl_2	reflux	15	36	38
3	4a	Ar	CH_2Cl_2	reflux	26	37	33
4	4a	Ar	CH_2Cl_2	reflux	48	33	31
5 ^b	4b	$CH_2 = CH_2$	Toluene	80	3	5	74
6	4b	$CH_2 = CH_2$	CH_2Cl_2	reflux	1.5	74 ^c	-

a) Yields were determined by ¹H NMR spectroscopy using (E)-stilbene as the internal standard.

b) 5 mol% of 1 was used.

c) Isolated yield.

Table 3.Synthesis of benzodiazepine derivatives.

Entry	Substrate	Solvent	Temp (°C)	Time (h)	7 (%) ^a
1	6a	Toluene	80	1	_
2	6a	CH_2Cl_2	reflux	28	-
3	6b	Toluene	80	0.5	99
4 ^b	6b	Toluene	80	0.5	99
5	6b	CH_2Cl_2	reflux	15	97
6 ^c	6b	CH_2CI_2	reflux	20	95

a) Isolated yield.

reaction time was required (Entry 5). Excellent yield of **7b** was also accomplished when the reaction was carried out under an argon atmosphere (Entry 6).

Construction of 8-Membered Ring

Encouraged by the success of the construction of a 7-membered ring, we were also interested in the synthesis of 8-membered heterocycles.

At first, to synthesize azocine derivative 9, various linear enynes were examined. However, cyclized product 9 was not obtained in any case (Scheme 1).

Thus, RCM of linear ene-ynamide 10, which contained two tosylamide groups in a chain, was examined. When a CH₂Cl₂ solution of 10a was carried out in the presence of 10 mol% of 1 upon heating for

21 h, starting material 10a was recovered in 33% yield (Table 4, Entry 1). The introduction of an ethoxycarbonyl group on the alkyne affected the formation of an 8-membered ring, and 11b could be obtained in 15% yield (Entry 2).^[7] Improved yield was accomplished when the reaction was performed under an argon atmosphere (Entry 3). Higher recovery of starting material 10b was observed by using toluene as the solvent (Entry 4). Successful result for the construction of an 8-membered ring could be achieved and 11b was obtained in 78% yield when the reaction was conducted under a highdilution condition (Entry 5).

The synthesis of benzodiazocine was examined. The reaction of **12a** with second-generation Grubbs catalyst **1** provided only recovered **12a** (Table 5, Entry 1), and the shorter reaction time did not affect the

$$R^{2} = H \text{ or } CO_{2}Et$$

$$\mathbf{8a}: R^{2} = H$$

$$\mathbf{8b}: R^{2} = CH_{2}OBn$$

$$\mathbf{8c}: R^{2} = -CH_{2}OCMe_{2}OCH_{2}-$$

Scheme 1.

Synthesis of azocine derivatives.

b) 5 mol% of 1 was used.

c) Reaction was carried out under argon atmosphere.

Table 4. Synthesis of diazocine derivatives.

TsN NTs
$$\begin{array}{c|c} R & 10 \text{ mol}\% \text{ 1} \\ \hline CH_2Cl_2, \text{ Reflux, Time} \\ \hline 10a: R = H \\ 10b: R = CO_2Et \\ \end{array}$$

Entry	Substrate	Atmosphere	Time (h)	11 (%) ^a	Recovery of 10 (%) ^a
1	10a	CH ₂ =CH ₂	21	-	33
2	10b	$CH_2 = CH_2$	21	15	49
3	10b	Ar	21	33	25
4 ^b	10b	$CH_2 = CH_2$	6	-	68
5°	10b	Ar	24	78 ^d	-

a) Yields were determined by ¹H NMR spectroscopy using (E)-stilbene as the internal standard.

result (Entry 2). However, when a reaction of **12b** was carried out using **1** upon heating in CH₂Cl₂ for 21 h, a new spot appeared on TLC (Entry 3). It seemed likely that 1 H NMR indicated that compound **13** had been obtained. However, the peaks were separated clearly when 1 H NMR was measured at a low temperature (-50 °C). Furthermore, the largest m/z value of FAB-MS spectra was at 1133 (M⁺ + H). These results indicated that the metathesis

product is dimeric compound **14**, not **13**. An argon atmosphere was effective for the dimerized cyclization of **12b**, and the yield of **14b** was increased to 52% (entry 4). Replacing the solvent to toluene from CH_2Cl_2 , a high recovery of **12b** was observed under an ethylene atmosphere (Entry 5), and a high yield of **14b** was provided under an argon atmosphere (Entry 6).

When metathesis product 14b was treated with $NaBH_4$ in the presence of

Table 5. Synthesis of benzodiazocine derivatives.

Entry	Substrate	Atmosphere	Time (h)	13 (%)	14 (%) ^a	Recovery of 12 (%) ^a
1.	12a	CH ₂ =CH ₂	21	-	-	58
2 ^b	12a	$CH_2 = CH_2$	0.5	-	_	53
3	12b	$CH_2 = CH_2$	21	-	14	63
4	12b	Ar	21	-	52	44
5. ^b	12b	$CH_2 = CH_2$	6	-	-	86
6 ^b	12b	Ar	2	-	77	-

a) Isolated yield.

 $^{^{}b)}$ Reaction was carried out in the presence of 5 mol% of 1 in toluene at 80 $^{\circ}$ C.

c) Reaction was carried out under low concentration condition (0.002 M).

d) Isolated yield.

b) Reactions were carried out in the presence of 5 mol% 1b in Toluene at 80 °C.

Scheme 2. 1,4-Reduction of **14b**.

NiCl₂, an exo olefin moiety of the metathesis product was reduced to provide **15b** (Scheme 2).

Conclusion

In summary, the synthesis of 7- and 8-membered heterocycles catalyzed by ruthenium carbene complex 1 was investigated. Various medium-sized heterocyclic compounds could be synthesized by RCM of ene-ynamides in high yields. In the case of RCM of 12b, 16-membered macrocycle 14b was produced.

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[4] The spectrum data of 2-(p-Toluenesulfonyl)-3-vinyl-2,5-dihydro-1H-benzo[c]azepine (3a): IR (neat) ν 1597 (w), 1338 (s), 1159 (s) cm $^{-1}$; 1H NMR (400MHz, CDCl $_3$) δ 2.30 (s, 3H), 3.30 (d, J=5.8Hz, 2H), 4.81 (s, 2H), 5.13 (d, J=10.6Hz, 1H), 5.43 (d, J=16.9Hz, 1H), 5.77 (t, J=5.8Hz, 1H), 6.38 (dd, J=10.6, 16.9Hz, 1H), 6.73 (d, J=7.7Hz, 1H), 6.97 (d, J=8.2Hz, 2H), 7.06 (ddd, J=1.9, 7.3, 7.3Hz, 1H), 7.11-7.19 (m, 2H), 7.27 (d, J=8.2Hz, 2H); 13 C NMR (100MHz, CDCl $_3$) δ 21.4, 33.9, 53.5, 115.1, 125.0, 126.3, 127.3, 127.4, 128.8, 129.0, 129.6, 129.8, 135.2, 135.3, 137.0, 141.1, 142.6; El-LRMS m/z 2325 (M $^+$), 260, 170, 117, 91; El-HRMS m/z calcd for $C_{19}H_{19}O_2NS$ (M $^+$) 325.1136, found 325.1121.

[5] The spectrum data of 1-(*p*-Toluenesulfonyl)-7-vinyl-2,3,4,5-tetrahydro-1*H*-azepine (**5a**): IR (neat) ν 1599 (w), 1343 (s), 1158 (s) cm⁻¹; ¹H NMR (400MHz, CDCl₃) δ 1.29-1.36 (m, 2H), 1.66-1.73 (m, 2H), 1.84-1.90 (m, 2H), 2.42 (s, 3H), 3.49 (br, 2H), 5.05 (d, J=10.6Hz, 1H), 5.29 (d, J=17.2Hz, 1H), 5.94 (dd, J=6.8, 7.3Hz, 1H), 6.25 (dd, J=10.6, 17.2Hz, 1H), 7.27 (d, J=8.2Hz, 2H), 7.76 (d, J=8.2Hz, 2H), ¹³C NMR (100MHz, CDCl₃) δ 21.5, 23.6, 26.0, 29.2, 49.1, 114.6, 127.5, 129.4, 131.9, 134.6, 139.1, 142.4, 143.1; El-LRMS m/z 277 (M⁺), 212, 155, 122, 91; El-HRMS m/z calcd for C₁₅H₁₉O₂NS (M⁺) 277.1136, found 277.1134.

[6] The spectrum data of Ethyl 2-[1,5-Bis-(p-toluenesulfonyl)-4,5-dihydro-1H-benzo[b][1,4]diazepin-2-yl]acrylate (**7b**): IR (KBr) v 1720 (m), 1598 (w), 1355 (s), 1164 (s) cm⁻¹; ¹H NMR (400MHz, CDCl₃) δ 1.24 (t, J=7.2Hz, 3H), 2.42 (s, 3H), 2.44 (s, 3H), 3.73 (dd, J=2.9, 18.4Hz, 1H), 4.06-4.15 (m, 2H), 4.57 (dd, *J*=3.9, 18.4Hz, 1H), 5.30 (dd, J=2.9, 3.9Hz, 1H), 5.88 (s, 1H), 6.25 (s, 1H), 7.26-7.36 (m, 7H), 7.46 (m, 1H), 7.72 (d, J=8.2Hz, 2H), 7.87 (d, J=8.2Hz, 2H); ¹³C NMR (100MHz, CDCl₃) δ 14.1, 21.6, 21.6, 49.1, 61.0, 121.4, 125.7, 127.5, 128.1, 128.3, 128.4, 129.1, 129.5, 129.9, 130.4, 136.3, 136.9, 137.0, 137.8, 139.4, 140.3, 143.7, 144.0, 165.1; EI-LRMS m/z 552 (M⁺), 507, 397, 242, 169, 139, 91; EI-HRMS *m/z* calcd for $C_{28}H_{28}O_6N_2S_2$ (M⁺) 552.1389, found 552.1399. [7] The spectrum data of Ethyl 2-[1,5-Bis-(toluene-4-sulfonyl)-1,4,5,6,7,8-hexahydro-[1,5]diazocin-2-yl]acrylate (11b): IR (neat) v 1717 (s), 1598 (w), 1343 (s), 1161 (s) cm⁻¹; ¹H NMR (400MHz, CDCl₃) δ 1.15 (t, J=7.3Hz, 3H), 1.86-1.93 (m, 2H), 2.40 (s, 3H), 2.44 (s, 3H), 3.41-3.47 (m, 2H), 3.63 (t, *J*=5.3Hz, 2H), 3.85 (d, *J*=8.2Hz, 2H), 3.90 (q, J=7.3Hz, 2H), 5.81 (s, 1H), 6.16 (s, 1H), 6.18 (t, J=8.2Hz,1H), 7.23 (d, J=7.7Hz, 2H), 7.31 (d, J=7.7Hz, 2H), 7.58 (d, J=8.2Hz, 2H), 7.65 (d, J=8.2Hz, 2H); ¹³C NMR (100MHz, $CDCl_3$) δ 13.9, 21.5, 21.5, 27.9, 44.1, 48.5, 51.8, 60.9, 127.0, 127.4, 129.4, 129.8, 130.5, 131.0, 136.2, 136.7, 137.8, 139.3, 143.4, 143.4, 165.0; EI-LRMS m/z 518 (M⁺), 473, 397, 363, 335, 317, 267, 238, 207, 180, 155, 134, 91; EI-HRMS m/z calcd for $C_{25}H_{30}O_6N_2S_2$ (M⁺) 518.1545, found 518.1538.